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SYNTHESIS OF 1-DEAZAADENOSINE ANALOGUES OF (2'→5') Apapa

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ABSTRACT: Synthesis of (2'→5')ApApA analogues containing 1-deazaadenosine at different positions is described (32-34). The approach used the phosphotriester methodology in solution and utilized 3'-O-benzoylated derivatives of the N^o-protected 5'-O-monomethoxytrityl-1-deazaadenosine as starting material.

INTRODUCTION

The 5'-triphosphates of $2'\rightarrow 5'$ -oligoadenylates $[ppp(A2'p)_n 5'A, n \ge 2;$ mainly trimers; $2'\rightarrow 5'A]$ are implicated at least in one of the mechanisms of antiviral actions of interferon (for review, see¹). The $2'\rightarrow 5'A$ binds to and subsequently activates a latent endoribonuclease (RNase L) resulting in the cleavage of virus mRNA and eventually in the inhibition of virus replication. A wide variety of $2'\rightarrow 5'A$ analogues

Dedicated to Prof. Y. Mizuno on the occasion of his 75th birthday

have been synthesized in order to establish the binding sites of the $2'\rightarrow 5'A$ molecule to RNase L. Thus, the replacement of adenosine by inosine²⁻⁴ or $1,N^6$ -ethenoadenosine⁵ resulted in dramatic decrease of biological activities. From these data, it has been reasoned that the adenine 6-amino group and possibly its N^1 nitrogen atom may be of crucial importance for the binding to and the activation of RNase $L^{3,4}$. In order to study the relative importance of nitrogen-1 as binding site of each of the adenine bases of $(2'\rightarrow 5')pppApApA$ in biochemical events the synthesis of those trimers was initiated in which one of the nucleotide fragment is sequentially replaced by 1-deazaadenosine (c^1A) . In this paper the synthesis of 1-deazaadenosine analogues (A-C) of $(2'\rightarrow 5')$ ApApA is described.

RESULTS AND DISCUSSION

Oligodeoxyribonucleotides containing 1-deaza-2'-deoxyadenosine have recently been prepared by solid-phase synthesis to generate Hoogsteen-duplex DNA⁶. The solid-phase oligoribonucleotides using P(III)-chemistry has also been established^{7,8}. For the synthesis of the $(2'\rightarrow 5')$ trimers phosphotriester methodology was chosen⁹.

NMR-Data and Glycosyl Bond Conformation of 1-Deazaadenosine.

1-Deazaadenosine (1a)^{10,11} was prepared either chemically by the very efficient method of Mizuno^{12,13} (cf. Ref. ¹⁴) or obtained by microbiological transglycosylation¹⁵. During the course of this work, various 6-substituted 1-deazapurine β-D-ribofuranosides, *viz.*, 6-nitro (1b), 6-chloro (1c)¹³, and 6-benzamido (2) were synthesized. All these compounds were characterized by ¹H- and ¹³C-NMR spectra (Tables 1, 2, and 3). The assignment of most resonances was made by 2D [¹H, ¹³C] correlation spectra and in doubtful cases, e.g., of OH-5' in c¹A, resolved by homo-decoupling experiments.

Systematic numbering; purine numbering in parenthesis.

TABLE 1. $^1\text{H-NMR}$ Chemical Shifts (δ), Multiplicities, and Coupling Constants (J) [Hz] of 1-Deazaadenosine Derivatives a).

Compd.	H-6 ^{b)} H-1 ^{c)}	H-5 ^{b)} H-2 ^{c)}	H-2 ^{b)} H-8 ^{c)}	H-1'	H-2'	H-3'	H-4'	H-5'/H-5''
1a	6.39d	7.78d	8.25s	5.87d	4.71m	4.13m	3.98m	3.67/3.56m
1b	8.04d	8.69d	9.11s	6.15d	4.61m	4.21m	4.00m	3.70/3.60m
1c	7.52d	8.36d	8.95s	6.09d	4.65m	4.21m	4.00m	3.70/3.62m
2*	7.89d	8.35d	8.69s	6.07d	4.72m	4.21m	4.02m	3.71/3.60m
2*	7.89d	8.35d	8.64s	6.08d	4.71g	4.20q	4.02m	3.71/3.61m
3*	7.80d	8.24d	8.60s	6.04d	4.69m	4.19m	4.00m	3.72/3.58dd
	6.40d	7.82d	7.85s	5.78d	5.52g	4.40d	4.36s	3.92/3.70m
4 5 6	5.75d	7.60d	8.11s	5.96d	4.70t	4.48m	4.32d	3.47/3.20dd
6	5.68d	7.52d	7.72s	5.74d	5.02m	4.26d	4.20s	3.82/3.52dd
7	8.28d	8.44d	8.22s	6.08d	4.84t	4.20d	4.27m	3.28/3.15m
8	8.20d	7.97d	8.23s	6.06d	4.80m	4.40m	4.45m	3.47/3.24dd
9	-	-	8.23s	6.68d	6.51t	6.15m	4.62m	3.61d
10*	8.12d	8.34d	8.66s	6.52d	6.18m	4.66m	4.26m	-
li.	-	8.30d	8.67s	6.22d	5.36m	5.73m	4.25m	-
12	-	-	-	6.64d	6.49t	6.11m	4.61m	3.62d
13	-	-	-	6.49d	6.13t	5.00m	4.34m	3.58/3.46dd
14 15* 16* 18*	-	-	-	6.17d	5.18t	5.70dd		3.55/3.38dd
15*	-	-	8.41s	5.64d	6.74m	6.18t	4.56m	-
16*	-	-	8.39s	6.08d	5.29m	5.64m	4.40m	-
18	-	8.24d	8.80s	6.42d	5.96m	5.80m	4.42m	-
23	-	-	-	6.06d	6.40m	6.40m	4.62s	4.05m
Compd.	δ and J(O	H-2')δ aı	nd J(OH-:	3')δ and	J(OH-5')	Ph	N.	H
1a	5.37 d (6	.4) 5.1	4 d (4.1)	6.06 s	1	_	6	.49 s
1b	5.59 d (5		7 d (4.6)	5.13 t		-	-	
1c	5.54 d (5		5 d (4.8)	5.19 t		-	-	
2*	5.53 d (6		0 d (4.7)	5.38 t		7.50-8.0		.48 s
2* 3*	5.63 d (6		5 d (4.7)		q (5.5)	7.54-8.0		.49 s
3	5.50 d (5	.2) 5.2	4 d (4.0)	5.41 s	3).13 s
7 *	-	-		-		7.72-8.0		.32 s
10*	-		2 d (6.0)	-		6.80-8.0		0.58 s
11.	6.01 d (6.			-		6.80-8.1		.52 s
16	5.98 d (6.	.5) -		-		6.76-8.0		
18 [™]	-	-		-		6.78-8.0	8 m 9	.90 s

 $^{^{}a)} Spectra were measured in CDCl_3 rel. to TMS; the compounds with asterisk were measured in DMSO-d_6.$ $<math display="inline">^{b)} Systematic numbering.$

TABLE 2. ${}^{1}_{H}$, ${}^{1}_{H}$ -Coupling Constants [Hz] and Chemical Shifts (δ) of 1-Deazaadenosine Derivatives a).

Compd.	J(1,2)	J(1',2')	J(2',3')	J(3',4')	J(4',5')	J(5',5")
1a	5.6	6.4	6.0	1.0		
1b	5.3	5.2	4.8	4.0		
1c	5.2	5.5	5.4	3.4		
2*	5.5 5.5 5.8	5.9	5.5 5.5	3.0 3.0	3.5	11.0
2*	3.3 5 0	5.9 6.0	5.2	3.0	3.5	12.0
<i>A</i>	5.5	7.5	5.0	< 1.0	<1.0	12.5
5	6.0	6.5	6.5	< 1.0	4.0	11.0
6	5.5	7.0	5.0	< 1.0	< 1.0	12.5
7	5.5	5.5	5.5	2.0	4.0	11.0
2* 2* 3 4 5 6 7 8 9 ** 10*	5.5 5.8	6.0	4.6	b)	4.0	11.0
9 *	b) 5.5 5.5	6.0	6.0	3.0	3.5	
10_*	5.5	4.0	5.0	5.5		
11	5.5	5.5	6.0	4.5	4.5	
12	b) b) b) b) b) 5.5	6.0	6.0	3.5 4.0	3.0 3.5	10.0
13	b)	5.0 6.2	5.0 6.0	2.0	3.3	11.0
14* 15	h)	5.5	5.0	5.0	4.0	11.0
14 _* 15 _* 16 _*	ь́	5.5 6.3	6.3	5.0	5.0	
18*	5.5	5.5	0.5	2.0	2.0	
23	b)	5.0		< 1.0	4.5	

a)See a) of Table 1. b)Not determined owing to overlap.

TABLE 3. $^{13}\text{C-NMR}$ Chemical Shifts (δ) of 1-Deazapurine β -D-Ribofuranosides $^{a,b)}$

Compd.	C-1 ^{b)} C-6 ^{c)}	C-2 ^{b)} C-5 ^{c)}	C-4 ^{b)} C-3a ^{c)}	C-5 ^{b)} C-7a ^{c)}	C-6 ^{b)} C-7 ^{c)}	C-8 ^{b)} C-2 ^{c)}
1a 1b 1g)	102.4 112.2 116.8 109.2	144.2 144.8 144.7 144.8	146.5 150.3 133.5 147.2	123.9 127.2 133.2 131.2	147.5 143.8 147.4 137.4	140.1 147.8 144.6 142.6
Compd.	C-1'	C-2'	C-3'	C-4'	C-5'	
1a 1b 1g)	88.7 88.0 87.8 88.0	72.9 73.9 73.6 73.9	71.2 70.0 70.3 70.6	86.3 85.5 85.6 85.7	62.2 61.0 61.2 61.6	

 $^{^{}a)}\mbox{In DMSO-d}_{6}$ rel. to TMS. $^{b)}\mbox{Purine numbering.}$ $^{c)}\mbox{Systematic numbering.}$ $^{d)}\mbox{In CDCl}_{3}$ rel. to TMS.

The ¹H-NMR-data for **1a** and **1c** correlate well with those previously published ^{13,16}. Assignments of quaternary ¹³C aglycone signals C-4, C-5, and C-6 (purine numbering) are based on the cumulative data for para-substituted pyridine derivatives ¹⁷ and are in excellent agreement with the data for related 2'-deoxy-β-D-ribofuranosides ¹⁸.

From Table 1 it is noteworthy that the OH-5' resonance signal of 1a is unusually deshielded and falls outside the range observed for other 6-substituted 1-deazapurine ribosides. This deshielding may result from an intramolecular 5'-OH···N³-hydrogen bonding and consequently from the *syn*-conformation at the glycosyl bond. The *syn*-conformation of compound 1a has already been detected by Mizuno et al. ¹⁶ using NOE-spectroscopy. We have also used ¹H-NMR NOE difference spectroscopy ¹⁹ for a more detailed conformational analysis with respect to the *syn/anti* equilibrium of different 1-deazapurine ribosides (Table 4).

As it can be seen the NOE data reveal a strong dependence of the N-glycosyl bond conformation upon the nature of the 6-substituent. A linear correlation of the *syn*-conformer population vs. the σ_{para} Hammett constants²⁰ for the substituents was found¹⁹.

According to that, the c^1A adopts about 95% syn conformation (cf. Ref. 16). Electron-donating 6-amino groups do facilitate the formation of intramolecular hydrogen bond between 5'-hydroxyl group and N^3 -nitrogen atom. There is also a good correlation between the NOE on H-2' upon saturation of H-8 (f_2 '8) as well as f_3 '8 and the syn/anti equilibrium. This coincides with the observation f_3 1 that the rather narrow syn region may be populated by the S-conformation of the furanose ring. It has been suggested that the syn conformation at the glycosyl bond is usually

TABLE 4. NOE Data of 1-Deazapurine-ß-D-ribofuranosides U	Jpon Irradiation of H-8. ^{a,b)}
--	--

Compd.	H-1'	NOE (%) H-2'	H-3'	Compd.	NOE (%) H-1'	H-2'	H-3'
1a	10.3	2.5	_	1c	5.9	4.5	1.3
1b	4.4	4.7	1.5	2	6.4	3.9	0.9

a)DMSO-d₆ at 23°C; b)Purine numbering.

accompanied by a C-2'-endo (S) pucker of the furanose ring (for a more detailed discussion, see Ref.²²). Indeed, the comparison of calculated²³ and experimental ³J[H,H] coupling constants of furanose ring protons clearly shows that within the series of 6-substituted 1-deazapurine ribosides, *viz.*, NO₂, Cl, NHBz, and NH₂, the population of C-2' *endo* (S) conformation is increased.

Protecting Groups.- In order to synthesize the c¹A-building blocks for the preparation of the desired trimers, various protecting groups for 6-amino residue were introduced: (i) Benzoylation of c¹A using the protocol of transient protection²⁴ afforded the benzoate 2.

(ii) In a similar way reaction with 2-(4-nitrophenyl)ethoxycarbonyl (NPEOC) chloride 25 gave compound 3. (iii) The dimethylaminomethylidene derivative 4 was prepared according to reference 26 . (iv) The reaction of c^1A with monomethoxytrityl chloride resulted in the formation of two compounds: the N^7 ,5'-O-bismonomethoxytrityl derivative 5 (70% yield) and the 5'-O-monoprotected 6 (15% yield). Monomethoxytritylation of 4 gave a complex reaction mixture the separation of which by silica gel column chromatography proved to be difficult and, therefore, 4 was not further investigated.

In order to make the correct choice for the deblocking conditions of the protected trimers, model studies with compounds 2, 3, and 5 were performed. Deprotection of 2 in conc. aq. ammonia for 20 h at room temperature failed. Treatment of 2 with conc. aqueous ammonia at 60°C for 6 h resulted in complete debenzoylation yielding

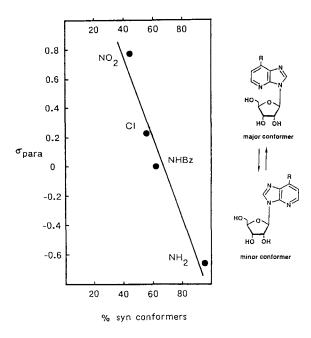


FIG. 1 Syn conformer populations of 1-deazapurine nucleosides as a function of Hammett constants of exocyclic substituents.

c¹A as the only reaction product. The half-life time of 2 with respect to deprotection under these conditions was found to be 125 min. The attempts of the deprotection of 2 with hydrazine hydrate in glacial acetic acid/pyridine²⁷ resulted in the formation of side products besides the expected c¹A that prevented this method of deblocking from further exploitation in oligomer synthesis. Next, treatment of the NPEOC derivative 3 with 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) in pyridine at room temperature for 16 h resulted in complete deprotection affording c¹A as the only reaction product (*cf.* Refs. 25, 28). Finally, treatment of the bis-protected compound 5 with 2% p-toluenesulfonic acid in a mixture of CH₂Cl₂/MeOH (7:3, v/v) at room temperature for 1.5 h gave quantitatively the inital nucleoside 1a.

Building Block and Oligonucleotide Synthesis.- It was previously shown that the 3'-O-benzoyl protection in combination with 5'-O-monomethoxytritylation and the 2-(4-nitrophenyl)ethyl (NPE) group for phosphate protection is effective for $(2'\rightarrow 5')$

oligonucleotide synthesis 9 ,29,30. The same strategy was studied for the synthesis of monomeric building blocks as well as the oligomers. Monomethoxytritylation of 2 and 3 gave the corresponding derivatives 7 and 8 in high yields. The method of selective 3'-O-benzoylation 9 using freshly destilled benzoyl chloride (1.2 eq.) in acetonitrile in the presence of Et₃N and DMAP was now carried out on compounds 7, 8, and 5 to give the respective 3'-O-benzoylated derivatives in the following yields: 11 (41%), 14 (58%), and 16 (53%). FAB mass spectra revealed the correct (M + H) $^+$ peaks (see Exp. Part).

Then compounds 11, 14, and 16 were transformed to the corresponding phosphotriesters 17, 19, and 21, and to the phosphodiesters 18, 20, and 22 under standard conditions³⁰. Benzoylation of 10 and subsequent detritylation gave the 2'-terminal c¹A building block 23. The synthesis of analogous adenosine building blocks 24a and 24b was described in ref.³⁰.

18 : R = Bz

20 : R = NPEOC

22 : R = MeOTr

	R	R ¹	\mathbb{R}^2	R ³	X
23		Bz	Bz	Bz	С
24a		Bz	Н	NPEOP(O)O	N
24b	н	Bz	Bz	Bz	N

		R ²	X	Y	Z
28	Bz	Bz Bz Bz NPEOC	С	N	N
29	H	Bz	С	N	N
30	Bz	Bz	N	N	С
31	Bz	NPEOC	N	С	N

The assembly of the trimers was performed by condensing (TPS-Cl/N-methylimidazole, 1:3, as an activating agent; CHCl₃ as a solvent^{31,32}) the monomeric building blocks in different successions and combinations in order to synthesize the 5'-detritylated dimers 25, 26 (see Exp. Part), and 27. Next, the dimers thus obtained were reacted with the phosphodiesters 18, 22, or 24a followed by detritylation to afford the partially blocked trimers 28-31. In the case of 29, detritylation (2% p-toluenesulfonic acid in CH₂Cl₂/MeOH, 7:3, v/v, at room

temperature for 2h) resulted in the formation of side products which seemed to be mainly chain-cleaved compounds³³.

Deprotection of **28** and **30** was performed by treatment with DBU followed by concentrated aqueous ammonia (40°C; 2h). Complex reaction mixtures were obtained from which the desired $2'\rightarrow 5'$ -trimers, $A_2(c^1A)$ (32) and $(c^1A)A_2$ (33) were isolated as Na⁺ salts in 8 and 17% yield.

An attempt to use the benzoate 18 for the synthesis of $A(c^1A)A$ (34) (condensation of 18 with 24b and the dimer obtained with 24a) resulted in the isolation of the trimer (34) in very low yield. Thus, exposure to concentrated ammonia at 40°C resulted in internucleotide cleavage (cf. ³⁴) especially in this latter case. Deprotection of 29 was performed by treatment with DBU, concentrated ammonia, and subsequent chromatography to afford the trimer $A_2(c^1A)$ (32) in 17% yield. Similar deprotection of 31 followed by chromatographic purification gave the trimer $A(c^1A)A$ (34) in 80% yield. The biological testing of the trimers 32-34 is in progress.

EXPERIMENTAL SECTION

General. Low resolution FAB mass spectra were obtained on a Kratos MS50TC (England) spectrometer from samples dissolved in DMSO with glycerol as matrix under Xe atoms bombardment (6-8 KeV). The UV-spectra were recorded on a Specord UV-VIS spectrophotometer (Carl Zeiss, Germany). 1 H-NMR spectra were recorded on a AC 250 spectrometer (Bruker, Germany) with tetramethylsilane as an internal standard (s = singlet; d = doublet; t = triplet; q = quartet; m = multiplet).

¹H-NMR spectra of trimers 32, 33, and 34 were measured in D₂O at 23 °C; the chemical shifts (δ) and coupling constants (J) are reported in ppm rel. to external TMS and Hz, respectively. NOE Spectra were measured on an AC-250 spectrometer (Bruker, Germany) as described in ref. ¹⁹. Thin layer chromatography (TLC) was carried out on silica gel F 1500 LS 254 plates (Schleicher & Schull, Germany). Solvent systems used: EtOAc-hexane, 1:1 (A); EtOAc-hexane, 7:5 (B); CHCl₃-MeOH, 24:1 (C); CHCl₃-MeOH, 19:1 (D); CHCl₃-MeOH, 9:1 (E); CHCl₃-MeOH, 4:1 (F); CHCl₃-MeOH-Et₃N, 9:0.2:0.8 (G); 2-PrOH-conc. NH₃-H₂O, 17:1:2 (H), n-BuOH-HOAc-H₂O, 5:3:2 (I). Column chromatography was performed on silica gel L 40/100 μ and 100/400 μ (Chemapol, Czechoslovakia). Melting points were determined with a Boethius (Germany) apparatus and are uncorrected. The solutions of compounds in organic solvents were dried with anhydr. Na₂SO₄ for 4h. The reactions were performed at room temperature, unless stated otherwiese. The nomenclature of oligonucleotides follows the (2'→5')-direction throughout the manuscript.

7-(Benzoylamino)-N³-(B-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (2). Trimethylsilyl chloride 10,11 (1.14 mL, 9.01 mmol) was added to a solution of 1-deazaadenosine 1a (0.32 g, 1.20 mmol) in anh. pyridine (3 mL) and the reaction mixture was stirred for 15 min. Benzoyl chloride (0.42 mL, 3.6 mmol) was then added and stirring was continued for 2h. The reaction mixture was cooled to 0°C and water (1.2 mL, 0°C) was added under stirring. After 5 min, concentrated aquaeous ammonia (2.4 mL) was added and stirring was continued for 30 min. The reaction mixture was evaporated and the residue was partitioned between water (30 mL) and CHCl₃ (30 mL). The organic layer was dried, evaporated, and the residue was crystallized from water to give colorless crystals (0.39 g, 87%); m.p. 228-230°C; R_f 0.4 (E); UV (MeOH) λ_{max} nm ($lg \epsilon$): 282 (4.35). FAB MS m/z 371 (M + H)⁺.

$7-\{[2-(4-Nitrophenyl)ethoxycarbonyl]amino\}-N^3-(\beta-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (3).$

Trimethylsilyl chloride (0.63 g, 5.8 mmol) was added to a solution of 1-deazaadenosine 1a, (0.22 g, 0.83 mmol) in anh. pyridine (2 mL) and the reaction mixture was stirred for 2h. Then, 2-(4-nitrophenyl)ethylchloroformate²⁵ (0.45 g, 1.96 mmol) was added and stirring was continued for 1h at 0°C and another 2h at room temperature. The reaction mixture was cooled to 0°C and water (0.78 mL, 0°C) was added under stirring. After 5 min, concentrated aqueous ammonia (1.56 ml) was added and stirring was continued for 30 min. The reaction mixture was

evaporated and the product was purified by silica gel column chromatography (70 ml). Elution was performed with a linear gradient (0 \rightarrow 10%, v/v, 600 mL) of MeOH in CHCl₃. The fractions containing the product were collected, evaporated, and crystallized from EtOH to give colorless crystals (0.29 g, 77%); m.p. 193-195°C; R_f 0.5 (D); UV (MeOH) λ_{max} nm (lg ϵ): 267 (4.47), 276sh (4.42). FAB MS m/z 460 (M + H)⁺.

7- $\{[(Dimethylamino)methylidene]amino\}-N^3-(\beta-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (4).$

To a solution of 1-deazaadenosine 1a (0.02 g, 0.075 mmol) in anh. DMF (0.4 ml), N,N-dimethylformamide dimethyl acetal (0.1 ml) was added and the mixture was kept for 4 days. The reaction mixture was evaporated, the residue was dissolved in CHCl₃ (0.2 mL), and precipitated in hexane (20 mL). The resulting precipitate was collected by filtration and dried *in vacuo* to give colorless 4 (0.02 g, 83%); m.p. 84-86°C, R_f 0.4 (F); UV (MeOH) λ_{max} nm (lg ϵ): 309 (4.54). FAB MS m/z 322 (M + H)⁺.

Reaction of 1-Deazaadenosine with Monomethoxytrityl Chloride.

A solution of 1-deazaadenosine 1a (0.11 g, 0.41 mmol), monomethoxytrityl chloride (0.21 g, 0.68 mmol) and 4-dimethylaminopyridine (6 mg, 0.052 mmol) in anh. pyridine (2 mL) was stirred for 12 h and then poured into a mixture of ice and water (150 mL) under vigorous stirring. The resulting precipitate was collected by filtration, washed with water, and dried *in vacuo*. The products (5 and 6) were purified on a silica gel (70 mL) column eluting with a linear gradient of methanol in EtOAc (0→5%, v/v; 1 L).

$\label{eq:continuous} 7-(Methoxytriphenylmethyl) - (\beta-D-ribofuranosyl)-3H-imidazo[4,5-b] pyridine~(5).$

Compound **5** (0.22 g, 70%) isolated as an amorphous powder; m.p. 131-133°C, R_f 0.6 (C). UV (MeoH) λ_{max} nm (lg ϵ): 270 (4.51), 287 (4.54). FAB MS m/z 811, 273 (M + H) $^+$.

7-Amino-N³-[5-*O*-(4-methoxytriphenylmethyl)-(β-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (6).

Compound **6** (0.03 g, 15%) isolated as an amorphous powder; m.p. 210-214°C; R_f 0.3 (D); UV (MeOH) λ_{max} nm (lg ϵ): 270 (3.32), 287 (3.41). FAB MS m/z 539, 273 (M + H)⁺.

$\label{eq:continuous} 7- (Benzoylamino)-N^3-[5-\emph{O}-(4-methoxytriphenylmethyl)-(\beta-D-ribofuranosyl]-3H-imidazo[4,5-b]pyridine~(7).$

A solution of 2 (0.35 g, 0.95 mmol) and 4-methoxytrityl chloride (0.38 g, 1.23 mmol) in anh. pyridine (7 mL) was stirred for 12 h and then poured into a mixture of ice and water (300 mL) under vigorous stirring. The resulting precipitate was collected by filtration, washed with water, and dried *in vacuo*. The product was purified on a silica gel column (100 mL) eluting with a linear MeOH gradient (0 \rightarrow 10%, v/v; 1 L) in CHCl₃ to give 7 as an amorphous powder (0.47 g, 79%; m.p. 108-111°C; R_f 0.4 (D). UV (MeOH) λ_{max} nm (lg ϵ): 230 (4.66), 280 (4.66). FAB MS m/z 643, 273, 369, 353 (M + H)⁺.

$\label{eq:continuous} 7-\{[2-(4-Nitrophenyl)ethoxycarbonyl]amino\}-N^3-[5-O-(4-methoxytriphenylmethyl)-(\beta-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine~(8)\,.$

Compound 8 was prepared as described above, starting from 0.24 g (0.52 mmol) of 3 to give a colorless solid (0.29 g, 78%); m.p. 102-105°C (EtOH); R_f 0.8 (D). UV (MeOH) λ_{max} nm (lg ϵ): 267 (4.56), 276sh (4.50). FAB MS m/z 732, 273 (M + H)⁺.

Benzoylation of 7-(Benzoylamino)- N^3 -[5-O-(4-methoxytriphenylmethyl)-(B-Dribofuranosyl)]-3H-imidazo[4,5-b]pyridine (7) with Benzoyl Chloride.

To the stirred solution of compound 7 (0.32 g, 0.5 mmol) in a mixture of anh. MeCN (7.5 mL), Et₃N (0.9 mL, 6.47 mmol) and 4-dimethylaminopyridine (DMAP) (4 mg, 0.04 mmol), freshly distilled benzoyl chloride (0.08 g, 0.069 mL, 0.6 mmol) was added. After stirring for 30 min, the reaction mixture was poured into a mixture of ice and water (200 mL) under vigorous stirring. After the ice was melted, the resulting precipitate was collected by filtration, washed with water, and dried *in vacuo*. The products (9-11) were chromatographed on a silica gel column (70 mL) eluting with a linear EtOAc gradient (20→60%, v/v; 500 mL) in hexane.

7-(Benzoylamino)- N^3 -[(2,3-O-dibenzoyl)-5-O-(4-methoxytriphenymethyl)]-(B-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (9).

The fractions containing **9** were pooled, evaporated, dissolved in EtOAc, and precipitated from hexane to give a solid (0.04 g, 10%); m.p. 94-98°C; R_f 0.3 (A); UV (MeOH) λ_{max} nm (lg ϵ): 230 (4.80), 279 (4.52).

7-(Benzoylamino)- N^3 -[(2-O-benzoyl)-5-O-(4-methoxytriphenylmethyl)-(B-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (10).

Compound 10 was isolated as colorless solid (0.06 g, 15%); m.p. 109-114°C; Rf 0.2

(A); UV (MeOH) λ_{max} nm (lg ϵ): 230 (4.64), 280 (4.45)]. FAB MS m/z 747, 273, 457 (M + H)⁺.

7-(Benzoylamino)- N^3 -[(3-O-benzoyl)-5-O-(4-methoxytriphenylmethyl)]-(B-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (11).

Compound 11 was obtained as colorless solid (0.15 g, 41%); m.p. 113-116°C; R_f 0.2 (A); UV (MeOH) λ_{max} nm (lg ϵ): 230 (4.64), 280 (4.45). FAB MS m/z 747, 273 (M + H)⁺.

Benzoylation of 7-{[2-(4-Nitrophenyl)ethoxycarbonyl]amino}-N³-{[5-0-(4-methoxytriphenylmethyl)]-(B-D-ribofuranosyl)}-3H-imidazo[4,5-b]pyridine (8). The reaction was performed as described above, starting from 0.22 g (0.30 mmol) of 8 giving compounds 12-14.

7-{[2-(4-Nitrophenyl)ethoxycarbonyl]amino}-N³-[(3-O-benzoyl)-5-O-(4-methoxytriphenylmethyl)-(B-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (14). Compound 14 was isolated as colorless solid (0.16 g, 58%); m.p. 106-108°C, R_f 0.3 (B); UV (MeOH) λ_{max} nm (lg ϵ): 232 (4.43), 267 (4.44), 276sh (4.39). FAB MS m/z 836 (M + H) $^+$.

7-{[2-(4-Nitrophenyl)ethoxycarbonyl]amino}-N³-[2-O-(benzoyl)-5-O-(4-methoxytriphenylmethyl)-(B-D-ribofuranosyl)]-3H-imidazo[4,5-b]pyridine (13). Colorless solid (30 mg, 11%); m.p. 109-116°C; R_f 0.4 (B); UV (MeOH) λ_{max} nm (lg ϵ): 232 (4.44), 267 (4.45), 277sh (4.40). FAB MS m/z 836, 273 (M + H) $^+$.

7-{[2-(4-Nitrophenyl)ethoxycarbonyl]amino}-N³-[2,3'-di-O-(benzoyl)-5-O-4-methoxytriphenylmethyl)-(B-D-ribofuranosyl)]-3H-imidazo[4,5-b]pyridine (12). Colorless solid (12 mg, 4%); m.p. 97-100°C; R_f 0.5 (B); UV (MeOH) λ_{max} nm (lg ϵ): 230 (4.63), 267 (4.51), 276sh (4.46. FAB MS m/z 940, 273 (M + H) $^+$.

Benzoylation of 7-(4-Methoxytriphenylmethylamino)-N³-[5-*O*-(4-methoxytriphenylmethyl)-(B-D-ribofuranosyl)]-3H-imidazo[4,5-b]pyridine (5). The reaction and workup was performed as described above, starting from 96 mg (0.12 mmol) of 5 yielding compounds 15 and 16 as amorphous solids.

7-(4-Methoxytriphenylmethylamino)-N³-[5-*O*-(4-methoxytriphenylmethyl)-3-*O*-(benzoyl)-(B-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (16).
Compound 16 was isolated as colorless solid (57 mg, 53%); m.p. 117-121°C; R_f 0.4

(MeOH) λ_{max} nm (lg ϵ): 230 (4.95), 277 (4.80).

(B); UV (MeOH) λ_{max} nm (lg ϵ): 270 (4.52), 287 (4.55). FAB MS m/z 915, 273 (M + H)⁺.

 $7-(4-Methoxytriphenylmethylamino)-N^3-[5-O-(4-methoxytriphenylmethyl)-2,3-di-O-(benzoyl)-(\beta-D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (15).$

Colorless solid (20 mg, 17%); m.p. 114-117°C; R_f 0.7 (B). FAB MS m/z 1019 (M + H) $^+$.

7-(Benzoylamino)- N^3 -{3-O-(benzoyl)-2-O-[2,5-dichlorophenyl-2-(4-nitrophenylethyl)-phosphato]-[5-O-(4-methoxytriphenylmethyl)]-(B-D-ribofuranosyl)}-3H-imidazo[4,5-b]pyridine (17).

Phosphorylation of **11** (0.15 g, 0.2 mmol) with 2,5-dichlorophenyldi(triazolido)phosphate followed by treatment with 2-(4-nitrophenyl)ethanol was carried out as described earlier 30 affording the triester **17** as an amorphous powder (0.15 g, 75%). R_f 0.2 (B); UV

 $\label{eq:continuous} $$7-\{[2-(4-Nitrophenyl)ethoxycarbonyl]amino}-N^3-\{2-O-[2,5-dichlorophenyl-2-(4-nitrophenylethyl)-phosphato]-[5-O-(4-methoxytriphenylmethyl)]-($B-D-ribofuranosyl)}-3H-imidazo[4,5-b]pyridine (19).$

Similarly, starting from 0.08 g (0.095 mmol) of **14**, 0.09 g (78%) of the triester **19** was obtained as an amorphous powder. R_f 0.3 (B); UV (MeOH) λ_{max} nm (lg ϵ): 274 (4.58).

7-(4-Methoxytriphenylmethylamino)- N^3 -{3-O-(benzoyl)-2-O-[2,5-dichlorophenyl-2-(4-nitrophenylethyl)-phosphato]-[5-O-(4-methoxytriphenylmethyl)]-(B-D-ribofuranosyl)}-3H-imidazo[4,5-b]pyridine (21).

Similarly, starting from 57 mg (0.062 mmol) of **16**, 64 mg (80%) of the triester **21** was obtained as an amorphous powder. R_f 0.4 (B); UV (MeOH) λ_{max} nm (lg ϵ): 270 (4.55), 287 (4.56).

 $7-(Benzoylamino)-N^3-\{3-O-(benzoyl)-5-O-(4-methoxytriphenylmethyl)-2-O-[2-(4-mitrophenylethyl)-phosphato]-(\beta-D-ribofuranosyl)\}-3H-imidazo[4,5-b]pyridine, Triethylammonium Salt (18).$

Treatment of the triester 17 (0.11 g, 0.058 mmol) with p-nitrobenzaloxime followed by work up as described earlier³⁰ afforded the diester 18 as an amorphous powder (0.09 g, 85%). R_f 0.3 (G); UV (MeOH λ_{max} nm (lg ϵ): 230 (4.81), 278 (4.77).

7-[2-(4-Nitrophenyl)ethoxycarbonylamino]-N³-{3-O-benzoyl-5-O-(4-methoxytriphenylmethyl)-2-O-[2-(4-nitrophenylethyl)-phosphato]-(B-D-ribofuranosyl)}-3H-imidazo[4,5-b]pyridine, Triethylammonium Salt (20). Similarly, starting from the triester 19 (0.09 g, 0.07 mmol), the diester 20 was obtained as an amorphous powder (0.064 g, 81%). Rf 0.3 (G); UV (MeOH) λ_{max} nm (lg ϵ): 267 (4.58), 276sh (4.56)].

7-(4-Methoxytriphenylmethylamino)- N^3 -{5-O-(4-methoxytriphenylmethyl)-2-O-[2-(4-nitrophenylethyl)-phosphato]-(B-D-ribofuranosyl)}-3H-imidazo[4,5-b]pyridine, Triethylammonium Salt (22).

In a similar way, starting from the triester **21** (64 mg, 0.05 mmol), the diester **22** was obtained as an amorphous powder (47 mg, 82%). R_f 0.4 (G); UV (MeOH) λ_{max} nm ($\lg \epsilon$): 270 (4.53), 287 (4.56).

7-(Dibenzoylamino)- N^3 -[2,3-O-(dibenzoyl)-(β -D-ribofuranosyl)-3H-imidazo[4,5-b]pyridine (23).

Benzoylation of **10** (0.124 g, 0.17 mmol) followed by detritylation as described earlier ³⁰ afforded a colorless solid (0.17 g, 66%); m.p. 190-191 °C (EtOH); R_f 0.4 (B); UV (MeOH) λ_{max} nm (lg ϵ): 230 (4.68), 270 (4.42). FAB MS m/z 683 (M + H) ⁺.

Synthesis of Dimers 25-27 and Trimers 28-31.

Synthesis of the dimers 25-27 and subsequently of the trimers 28-31 was performed by standard methodology consisting of (i) condensation of phosphodiesters 18, 20, or 24a with 2'-terminal building blocks 23 or 24b and subsequent detritylation, and (ii) condensation of individual dimers obtained with phosphodiesters 18, 20, 22, or 24a and subsequent detritylation. To a solution of the appropriate phosphodiester (0.11 mmol) and the 2'-terminal building block or dimer (0.1 mmol) in CHCl₃ (0.5 mL), 2,4,6-triisopropylbenzenesulfonyl chloride (0.3 mmol) and N-methylimidazole (0.9 mmol) were added and the reaction mixture was stirred for 30 min. The reaction mixture was poured into hexane (200 mL), the resulting precipitate was collected by filtration, dried *in vacuo*, and then dissolved in 2% solution of p-toluenesulfonic acid in CH₂Cl₂/MeOH (7:3, v/v; 15 mL). After stirring for 5 min (in the case of 29, reaction time was 120 min), the solution was diluted with CHCl₃ (15 mL) and washed with 0.05 M phosphate buffer, pH 7.0 (2 x 30 mL). The organic layer was separated, dried, evaporated, and purified by silica gel column chromatography (60 mL). The product was eluted with a linear methanol gradient (0-5%, v/v; 2 x 300

mL) in chloroform. Appropriate fractions were collected, evaporated to a volume of 2 mL and precipitated into hexane (200 mL). The compounds were obtained as amorphous solids.

25: 71%; R_f 0.35 (C); UV (MeOH) λ_{max} nm (lg ϵ): 232sh (4.86), 276 (4.76).

26: 96%; R_f 0.34 (C); UV (MeOH) λ_{max} nm (lg ϵ): 267 (4.66), 277sh (4.63).

28: 70%; R_f 0.22 (C); UV (MeOH) λ_{max} nm (lg ϵ): 233 (4.89), 277 (4.79).

29: 66%; R_f 0.19 (C); UV (MeOH) λ_{max} nm (lg ϵ): 232 (4.84), 276 (4.72).

30: 67%; R_f 0.23 (C); UV (MeOH) λ_{max} nm (lg ϵ): 232sh (4.96), 276 (4.82).

31: 90%; R_f 0.21 (C); UV (MeOH) λ_{max} nm (lg ϵ): 267 (4.75), 277sh (4.74).

Deprotection of the trimers 28-31 was performed by the sequence deblocking of the phosphate group and subsequent treatment with methanol, saturated with ammonia at 0°C or conc. aqueous ammonia.

 $(2'\rightarrow 5')$ -Adenylyl- $(2'\rightarrow 5')$ -adenylyl- $(2'\rightarrow 5')$ -1-deazaadenosine, Sodium Salt (32).

Procedure a: The trimer **28** (0.11 g, 0.054 mmol) was dissolved in 0.5 M solution of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) in pyridine (16 mL) and stirred for 2h. After the addition of 1 M solution of HOAc in pyridine (8 mL), the mixture was evaporated and then coevaporated with pyridine (2 x 5 mL). The residue was dissolved in conc. ammonia (25 mL), kept at 40°C for 2h and evaporated. The residue was chromatographed on a DEAE-Sephadex A-25 (HCO3⁻ form, 50 mL) column using a linear gradient of aqueous triethylammonium hydrogen carbonate (0.001 \rightarrow 0.6 M; 500 mL, each). The product-containing fractions were collected, evaporated and precipitated as Na ⁺ salt³⁵ to give the trimer **32** (5 mg, 8%). R_f 0.3 (H); 0.07 (I); hypochromicity: 14%. ¹H-NMR (D₂O, purine numbering): 8.13, 8.05, 7.93, 7.90, and 7.72 (s, 5H, adenine H-8 and H-2, and H-8 of c¹A), 7.31 (d, 1H, J = 6.0 Hz, H-2 of c¹A), 6.03 (d, 1H, J = 4.5 Hz, H-1' of Np-), 5.96 (d, 1H, J = 4.5 Hz, H-1' of -pN).

Procedure B: Compound **29** (60 mg, 0.031 mmol) was treated with DBU in pyridine (9 mL, 2h) and then with saturated methanolic ammonia (15 mL, 20h); isolation as above afforded the trimer **32** (6 mg, 17%).

1-Deazaadenylyl-(2'→5')-adenylyl-(2'→5')-adenosine, Sodium Salt (33).

In a similar way, compound **30** (0.17 g, 0.08 mmol) was treated with DBU in pyridine (30 ml, 2h) and then with conc. ammonia (30 ml, 50 °C, 5h); chromatography followed by precipitation as above gave the trimer **33** (16 mg, 17%). R_f 0.3 (H); 0.07 (I); hypochromicity: 22%. 1 H-NMR (D₂O, purine numbering):

8.09, 7.99 (2H), 7.87, and 7.69 (s, 5H, adenine H-8 and H-2, and H-8 of c^1A), 7.87 (d, 1H, J = 6.0 Hz, H-2 of c^1A), 6.54 (d, 1H, J = 6.5 Hz, H-1 of c^1A), 6.08 (d, 1H, J = 4.0 Hz, H-1' of Np-), 5.94 (d, 1H, J = 3.5 Hz, H-1' of -pNp-), 5.88 (d, 1H, J = 4.5 Hz, H-1' of -pN).

Adenylyl-(2'→5')-1-deazaadenylyl-(2'→5')-adenosine, Sodium Salt (34).

Compound **31** (55 mg, 0.026 mmol) was treated with DBU in pyridine (20 mL) for 16h and worked up further as described for **29** to give the trimer **34** (23 mg, 80%). Rf 0.3 (H); 0.07 (I); hypochromicity: 27%. ¹H-NMR (D₂O, purine numbering): 8.16, 8.06, 8.00, 7.94, and 7.69 (s, 5H, adenine H-8 and H-2, and H-8 of c^1A), 7.64 (d, 1H, J = 6.0 Hz, H-2 of c^1A), 6.18 (d, 1H, J = 6.0 Hz, H-1 of c^1A), 6.07 (d, 1H, J = 4.0 Hz, H-1' of Np-), 6.00 (d, 1H, J = 3.5 Hz, H-1' of -pNp-), 5.78 (d, 1H, J = 3.5 Hz, H-1' of -pN).

Determination of Hypochromicity. To a solution containing 0.05 M Tris-HCl, 5 mM MgCl₂ and a specific amount of the trimer in the form of Na-salt (1 mL, optical density within 0.4-0.5; pH 8.8), the solution (50 μ L) of snake venom phosphodiesterase (1 μ g; Boehringer Mannheim, Germany) in the same buffer was added and the reaction mixture was incubated at 38°C until the absorbance reached the constant value. Hypochromicity was calculated as described in ref.³⁶.

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REFERENCES

- Torrence, P.F. in *Biological response modifiers. New approaches to disease intervention*, (Torrence, P.F. ed.), 1985, pp. 77-105, Academic Press, Orlando.
- Torrence, P.F.; Imai, J.; Lesiak, K.; Jamoulle, J.-C.; Sawai, H. J. Med. Chem. 1984, 27, 726.
- 3. Imai, J.; Lesiak, K.; Torrence, P.F. J. Biol. Chem. 1985, 260, 1390.
- 4. Torrence, P.F.; Imai, J.; Jamoulle, J.-C.; Lesiak, K. Chem. Scr. 1986, 26, 191.
- 5. Lesiak, K.; Torrence, P.F. FEBS Lett. 1983, 151, 291.

- 6. Seela, F.; Wenzel, T. Helv. Chim. Acta 1994, 77, 1485.
- 7. Seela, F.; Wenzel, T.; Debelak, H. Nucleosides & Nucleotides 1995, 14, 957.
- 8 Seela, F.; Debelak, H.; Rosemeyer, H.; Thomas, H.; Wenzel, T.; Zulauf, M. *Nucleic Acids Research Symp. Series No. 31* **1994**, 151.
- 9. Kalinichenko, E.N.; Podkopaeva, T.L.; Poopeiko, N.E.; Kelve, M.; Saarma, M.; Mikhailopoulo, I.A.; van den Boogaart, J.E.; Altona, C. Recl. Trav. Chim. Pays-Bas 1995, 114, 43.
- 10. Jain, S.K.; Chatterjee, S.K.; Ann, N. Indian J. Chem. 1966, 4, 403.
- 11. De Ross, K.B.; Salemink, C.A. Recl. Trav. Chim. Pays-Bas 1971, 90, 654.
- 12. Itoh, T.; Kitano, S.; Mizuno, Y. J. Heterocycl. Chem. 1972, 9, 465.
- 13. Itoh, T.; Sugawara, T.; Mizuno, Y. Nucleosides & Nucleotides 1982, 1, 179.
- Cristalli, G.; Franchetti, P.; Grifantini, M.; Vittori, S.; Bordoni, T.; Geroni,
 C. J. Med. Chem. 1987, 30, 1686.
- Mikhailopoulo, I.A.; Zinchenko, A.I.; Bokut, S.B.; Dudchik, N.V.; Baraj,
 V.N.; Kalinichenko, E.N.; Rosemeyer, H.; Seela, F. *Biotechnol. Lett.* 1992,
 14, 885.
- 16. Kitano, S.; Mizuno, Y.; Ueyama, M.; Tori, K.; Kamisaku, M.; Ajisaka, K. Biochem. Biophys. Res. Commun. 1975, 64, 996.
- 17. Kalinowski, H.-O.; Berger, S.; Braun, S. ¹³C-NMR Spektroskopie, G. Thieme Verlag, Stuttgart-New York, 1984, pp. 396-397.
- 18. Seela, F.; Wenzel, T. Heterocycles 1993, 36, 237.
- Rosemeyer, H.; Toth, G.; Golankiewicz, B.; Kazimierczuk, Z.; Bourgeois,
 W.; Kretschmer, U.; Muth, H.-P.; Seela, F. J. Org. Chem. 1990, 55, 5784.
- 20. Swain, C.G.; Lupton Jr., E.C. J. Am. Chem. Soc. 1968, 90, 4328.
- 21. Lüdemann, H.-D.; Roeder, O.; Westhof, E.; von Goldammer, E.; Mueller, A. *Biophys. Struct. Mechanism* 1975, 1, 121.
- 22. Birnbaum, G.I.; Shugar, D. Biologically active nucleosides and nucleotides: conformational features and interactions with enzymes. In: Topics in Molecular and Structural Biology, vol. 9 (Topics in Nucleic Acid Structure, Part 3; Neidle, S. ed.) pp. 10-23; The Macmillan Press Ltd., 1987.
- 23. de Leeuw, F.A.A.M.; Altona, C. J. Chem. Soc. Perkin II 1982, 375.
- 24. Ti, G.S.; Gaffney, B.L.; Jones, R.A. J. Am. Chem. Soc. 1982, 104, 1316.
- 25. Himmelsbach, F.; Schulz, B.S.; Trichtinger, T.; Charubala, R.; Pfleiderer, W. *Tetrahedron* **1984**, *40*, 59.
- 26. Holy, A.; Zemlicka, J. Coll. Czech. Chem. Commun. 1969, 34, 2449.
- 27. Letsinger, R.L.; Miller, P.S.; Grams, D.M. Tetrahedron Lett. 1968, 2621.

- 28. Charubala, R.; Uhlmann, E.; Himmelsbach, F.; Pfleiderer, W. Helv. Chim. Acta 1987, 70, 2028.
- 29. Kvasyuk, E.I.; Kulak, T.I.; Zaitseva, G.V.; Mikhailopoulo, I.A.; Charubala, R.; Pfleiderer, W. *Tetrahedron Lett.* **1984**, *25*, 3683.
- 30. Kvasyuk, E.I.; Kulak, T.I.; Khripach, N.B.; Mikhailopoulo, I.A.; Uhlmann, E.; Charubala, R.; Pfleiderer, W. *Synthesis* **1987**, 535.
- 31. Efimov, V.A.; Reverdatto, S.V.; Chakhmakcheva, O.G. *Bioorgan. Khim.* (*Moscow*) **1982**, 8, 231.
- 32. Zarytova, V.F.; Ivanova, E.M.; Romanenko, V.P. *Bioorgan. Khim. (Moscow)* 1983, 9, 516.
- 33. Tanimura, H.; Imada, T. Chem. Lett. 1990, 2081.
- 34. Kempe, T.; Chow, F.; Sundquist, W.I.; Nardi, T.J.; Paulson, B.; Peterson, S.M. *Nucl. Acids Res.* **1982**, *10*, 6695.
- 35. Moffatt, J.G. Can. J. Chem. 1964, 42, 599.
- 36. Huss, S.; Gosselin, G.; Pompon, A.; Imbach, J.-L. *Nucleosides & Nucleotides* 1986, 5, 275.